

Name: \_\_\_\_\_

Section: \_\_\_\_\_

## Report Form – Gas Stoichiometry

## Part I – Sample Data for Mass of Zinc

Chemical Reaction		
<b>DATA COLLECTED</b>		
Volume of hydrogen collected*	81.5 mL	L
Temperature of hydrogen*	22.0 °C	K
Barometric pressure*	29.98 in Hg	mm Hg
Height of solution in eudiometer from benchtop	19.2 cm	
Height of solution in beaker from benchtop	10.0 cm	
<b>CALCULATIONS AND RESULTS</b>		
Difference in liquid levels of solution in eudiometer and beaker*		
Aqueous vapor pressure at temperature of hydrogen	mm Hg	
Pressure caused by acid column:*( (Difference in cm)*(0.772 mm Hg/cm)	mm Hg	
Pressure of hydrogen alone*	mm Hg	atm
Moles of hydrogen*	moles	
Moles of zinc*	moles	
Mass of zinc (calculated)*	g	

Show the calculations for each of the entries in the Data Table marked with \* on the calculations page.

**Question:** If the mass of zinc used was 0.21 g, what is the percent error for your calculated mass of zinc? Show your work below.

Name: \_\_\_\_\_

Section: \_\_\_\_\_

**Part II – Mass of Magnesium**

Chemical Reaction		
<b>DATA COLLECTED</b>		
Unknown number		
Volume of hydrogen collected*	mL	L
Temperature of hydrogen*	°C	K
Barometric pressure*	inches Hg	mm Hg
Height of solution in eudiometer from benchtop	cm	
Height of solution in beaker from benchtop	cm	
<b>CALCULATIONS AND RESULTS</b>		
Difference in liquid levels of solution in eudiometer and beaker*	cm Acid Solution	
Aqueous vapor pressure at temperature of hydrogen	mm Hg	
Pressure caused by acid column:*( (Difference in cm) * (0.772 mmHg/cm)	mm Hg	
Pressure of hydrogen alone*	mm Hg	atm
Moles of hydrogen*	moles	
Moles of magnesium*	moles	
Mass of magnesium*	g	

Show the calculations for each of the entries in the Data Table marked with \* on the calculations page.

### Calculations

Show the calculations for each of the entries in the Data Table marked with \* below.

**VAPOR PRESSURE OF WATER AT DIFFERENT TEMPERATURES**

Temp (°C)	Vapor Pres (mm Hg)	Temp (°C)	Vapor Pres (mm Hg)
0	4.6	25	23.5
5	6.5	26	25.2
10	9.2	27	26.7
15	12.8	28	28.3
16	13.6	29	30.0
17	14.5	30	31.8
18	15.5	35	42.2
19	16.5	40	55.3
20	17.5	45	71.9
21	18.6	50	92.5
22	19.8	60	149.4
23	21.1	70	233.7
24	22.4	100	760.0